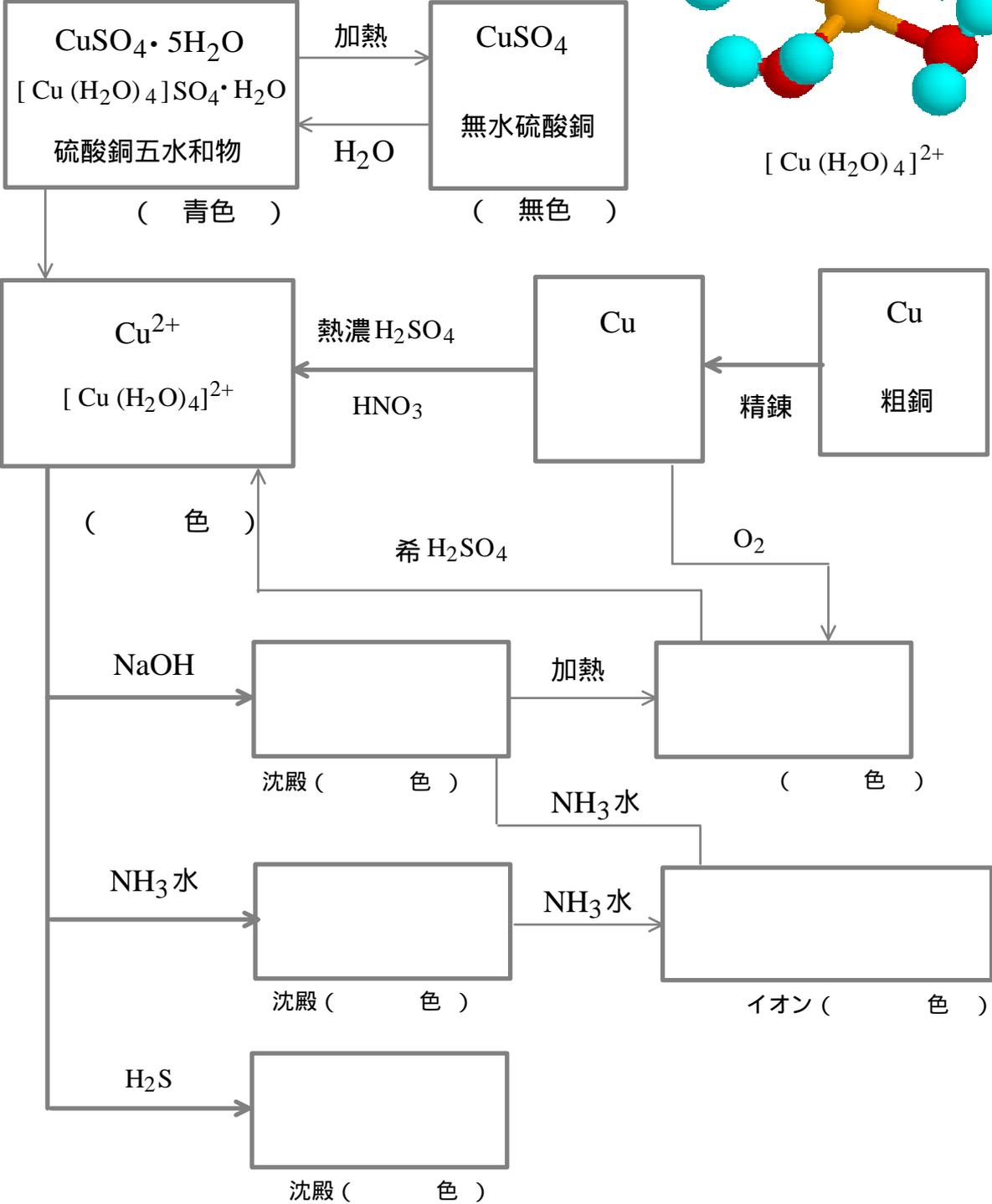
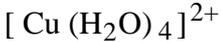
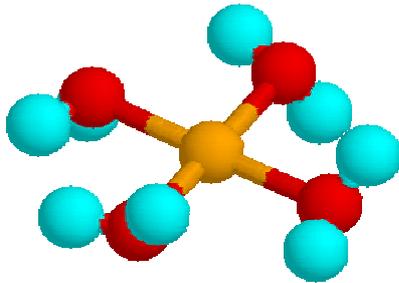


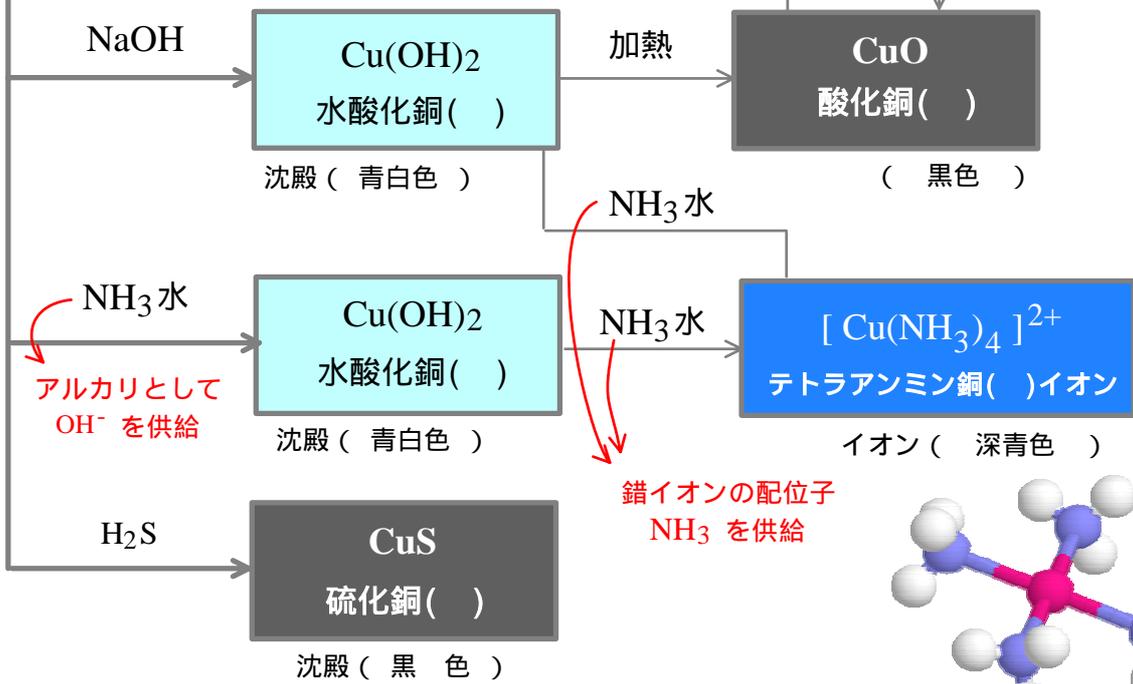
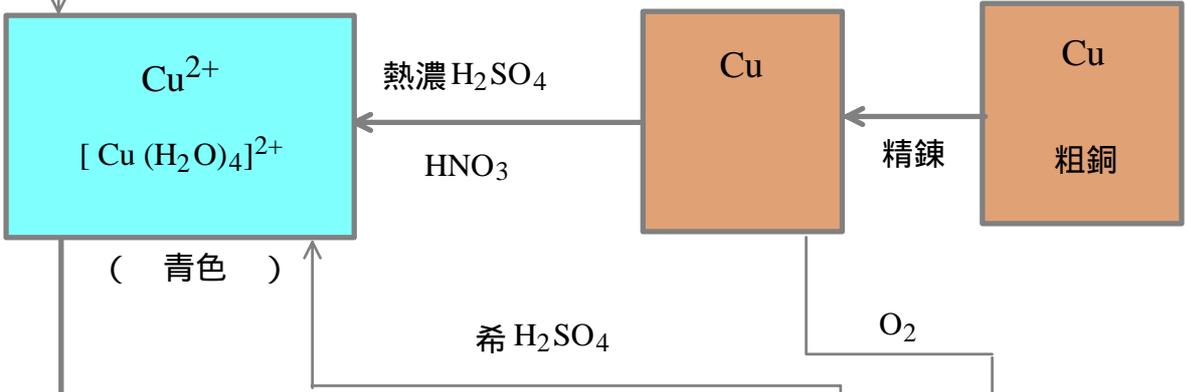
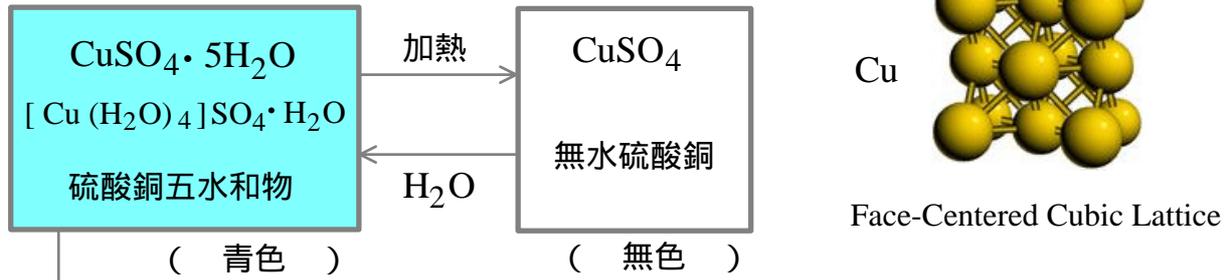
銅の変化



濃硝酸

希硝酸

銅の変化



	$\text{Cu} + 2\text{H}_2\text{SO}_4$	$\text{CuSO}_4 + 2\text{H}_2\text{O} + \text{SO}_2$
濃硝酸	$\text{Cu} + 4\text{HNO}_3$	$\text{Cu}(\text{NO}_3)_2 + 2\text{H}_2\text{O} + 2\text{NO}_2$
希硝酸	$3\text{Cu} + 8\text{HNO}_3$	$3\text{Cu}(\text{NO}_3)_2 + 4\text{H}_2\text{O} + 2\text{NO}$
	$\text{CuO} + \text{H}_2\text{SO}_4$	$\text{CuSO}_4 + \text{H}_2\text{O}$
	$\text{Cu}(\text{OH})_2 + 4\text{NH}_3$	$[\text{Cu}(\text{NH}_3)_4]^{2+} + 2\text{OH}^-$